1. (45 points) A $300 \mathrm{~kg} / \mathrm{hr}$ feed stream with composition 42 mass \% solute (C) and 58 mass \% water (diluent $A$ ) is to be contacted with solvent (B) in a countercurrent liquid extraction battery. Entering solvent is pure. The exiting raffinate should contain 16 mass \% solute ( $C$ ) on a solvent free basis. What is the minimum solvent flow rate required to achieve the desired composition of the exiting raffinate (corresponding to an infinite number of stages). A phase diagram is provided.

Be sure to label points $\Delta_{\text {min }}, V_{N+1}, V_{1 \text { min }}, L_{N}, L_{N}$, and $\mathrm{L}_{\mathbf{0}}$
2. (45 points) Feed to a leaching process consists of $1 \mathrm{~kg} / \mathrm{min} \mathrm{CaCO}_{3}$ (insoluble matrix) which carries in its pores $0.4 \mathrm{~kg} / \mathrm{min} \mathrm{NaOH}$ (solute) and $0.6 \mathrm{~kg} / \mathrm{min} \mathrm{H}_{2} \mathrm{O}$ (solvent). The entering solvent stream is $3 \mathrm{~kg} / \mathrm{min}$ of pure $\mathrm{H}_{2} \mathrm{O}$. Retention of solution by the $\mathrm{CaCO}_{3}$ is given by the following table. Leaching is carried out with a mixer-settler equivalent to a single equilibrium stage.
a. Calculate the NaOH mass fraction of the strong solution (exiting solvent stream with leached NaOH ).
b. What is the percent recovery of the NaOH ?

| NaOH, wt \% | 0 | 5 | 10 | 15 | 20 |
| :--- | :---: | :---: | :---: | :---: | :---: |
| kg solution/kg CaCO | 3 | 1.5 | 1.75 | 2.20 | 2.70 |

3. ( 10 points) The following mixture will be separated via a train of fractionating columns:

| Component | Mole Fraction | Boiling Point (C) |
| :---: | :---: | :---: |
| A | 0.20 | 120 |
| B | 0.20 | 125 |
| C | 0.20 | 130 |
| D | 0.20 | 140 |
| E | 0.20 | 190 |

If the first column is designed with component $\mathbf{B}$ as the light key and component $\mathbf{E}$ as the heavy key, where do the various components exit the first column? Complete the following table describing what fraction of the distillate and bottoms are composed of each component:

| Component | Presence in Distillate | Presence in Bottoms |
| :---: | :---: | :---: |
| A |  |  |
| B |  |  |
| C |  |  |
| D |  |  |
| E |  |  |

Label Product mole fraction as: large / small / negligible relative to the feed mole fraction. Negligible means that component is almost undetectable.


Phase Diagram for Problem 1

## Tie lines

| $(0.001,0.08)$ and $(0.753,0.18)$ | $(0.019,0.36)$ and $(0.535,0.42)$ |
| :--- | :--- |
| $(0.006,0.17)$ and $(0.67,0.27)$ | $(0.024,0.42)$ and $(0.41,0.53)$ |
| $(0.01,0.25)$ and $(0.62,0.33)$ | $(0.04,0.475)$ and $(0.35,0.58)$ |
| $(0.015,0.31)$ and $(0.573,0.38)$ | $(0.052,0.52)$ and $(0.273,0.62)$ |

